



**Directions:** *Unscramble the terms in italics to complete the sentences below. Write the terms on the lines provided.*

- \_\_\_\_\_ 1. The number of protons plus the number of neutrons gives the atomic *sams*.
- \_\_\_\_\_ 2. An atomic particle **NOT** in the nucleus is a(n) *roltecen*.
- \_\_\_\_\_ 3. A(n) *meeteln* is a material that contains only one kind of atom.
- \_\_\_\_\_ 4. An element that has characteristics of both metals and nonmetals is a *tademlilo*.
- \_\_\_\_\_ 5. A(n) *dunomcop* is a pure substance whose smallest unit is made of atoms of more than one element.
- \_\_\_\_\_ 6. An atomic particle with no electrical charge is a(n) *ennrout*.
- \_\_\_\_\_ 7. Two or more substances form a *tumirex* when they come together without forming a new substance.
- \_\_\_\_\_ 8. The number of protons in the *sculune* of an atom is the atom's atomic number.
- \_\_\_\_\_ 9. Elements that are *selamt* are usually shiny and conduct electricity well.
- \_\_\_\_\_ 10. Anything that takes up space and has mass is *tramte*.
- \_\_\_\_\_ 11. A *tronop* is a positively charged particle in the nucleus.
- \_\_\_\_\_ 12. Matter that has the same composition and properties throughout is called a *sceuntabs*.
- \_\_\_\_\_ 13. Atoms of the same element that have different numbers of neutrons are called *psoitoes*.
- \_\_\_\_\_ 14. An element's *coatim* number tells you the number of protons in its nucleus.